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SECOND OCCURRENCE OF PITIGLIANOITE, A MINERAL OF THE CANCRINITE-GROUP

Abstract - Pitiglianoite is a rare feldspathoid belonging to the cancrinite-group, originally found at Pitigliano, Southern Tuscany, Italy. The second occurrence of this minerals from another locality (Monte Cavalluccio, locality of Campagnano, province of Rome) is described. The cell parameters of the mineral are a=22.230(1), c=5.2679(5) Å, space group $P6_3$; its chemical composition is $Na_5K_3Si_6Al_6O_{24}(SO_4)^2H_2O$.

Key words - Pitiglianoite, cancrinite-group, feldspathoids.

Riassunto - Secondo ritrovamento della pitiglianoite, un minerale del gruppo della cancrinite. La pitiglianoite e' un raro feldspatoide appartenente al gruppo della cancrinite. E' stata trovata per la prima volta a Pitigliano, Toscana meridionale, Italia. Viene qui descritto il secondo ritrovamento di questo minerale, all'interno di un proietto vulcanico raccolto sul Monte Cavalluccio, in località Campagnano, provincia di Roma. I parametri di cella, ottenuti al diffrattometro automatico a quattro cerchi, sono a = 22.230(1), c = 5.2679(5) Å, gruppo spaziale $P6_3$; la composizione chimica del minerale è $Na_5K_3Si_6Al_6O_{24}(SO_4)$ °2H.O.

Parole chiave - Pitiglianoite, gruppo della cancrinite, feld-spatoidi.

INTRODUZIONE

Pitiglianoite was firstly found within ejected metasomatic blocks in a pumice quarry, near Pitigliano, southern Tuscany, Italy (Leoni et al., 1979). Originally, it was named «microsommite from Pitigliano», because its cell parameters closely resemble those of microsommite, a feldspathoid from Vesuvius (Bariand et al., 1968); however, recent chemical and structural data (Merlino et al., 1991) showed that pitiglianoite is a distinct mineralogical species. The genesis of this mineral is related to the interaction of a trachitic magma with the sulphate rocks of the volcanic vent.

In this paper we describe the second occurrence of pitiglianoite, which was found within a sanidinerich volcanic ejectum near Monte Cavalluccio, Campagnano, province of Rome, Italy. Here, pitiglianoite occurs as prismatic, elongated [0001], vitreous transparent crystals, about 1 mm long and 0.2 mm in diameter (Fig. 1). Associated minerals are sanidine, aegirine-augite, titanite, zircon, magnetite, nepheline and a hellandite-group mineral. In the same hand specimen, the new mineral peprossiite-

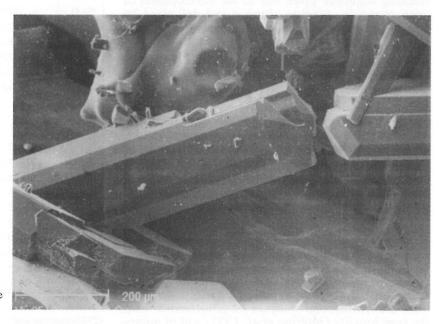


Fig. 1 - SEM microphoto of a pitiglianoite crystal in the sample from Campagnano.

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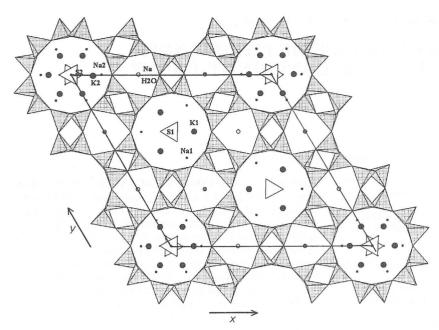


Fig. 2 - Schematic drawing of the structure of pitiglianoite as seen along [001]. A disordered distribution of the alkali cations and sulphate groups occurs in the channels at (0, 0, z) whereas the channels at (1/3, 2/3, z) and (2/3, 1/3, z) are ordered.

Ce (Della Ventura et al., 1993) was identified for the first time.

GENERAL FEATURES

Both microsommite and pitiglianoite belong to the cancrinite-group of minerals, which are characterized by the ABAB ... stacking along c of six-membered rings of [SiO₄] and [AlO₄] tetrahedra. This stacking sequence gives rise to the development of open channels, running along c and delimited by columns of undecahedral cages (Fig. 2). If we consider the type of atoms and molecules that occupy the undecahedral cages, we can classify these minerals in two subgroups: cancrinite-type minerals, with Na and H₂O (cancrinite, vishnevite, pitiglianoite), and davyne-type minerals, with Ca and CI (davyne, microsommite, quadridavyne). The large channels may contain several kinds of anions (CO₃²⁻, SO₄²⁻, Cl⁻, OH), cations (Na+, K+, Ca2+) and H2O molecules. In pitiglianoite, as well as in microsommite, an ordered distribution of sodium and potassium cations, and sulphate groups occurs, giving rise to a three-fold supercell with respect to those of cancrinite and davyne, respectively, with $A = \sqrt{3} a$ and C = c, where a and c are the cell parameters of cancrinite and davyne.

CHEMICAL DATA

The chemical analysis of the sample of pitiglianoite is reported in the first column of Table 1, together with the microprobe analyses of pitiglianoite from the type locality (Merlino *et al.* 1991) and of microsommite from Vesuvius. The chemical data were obtained with an EDS device attached to a Philips

Tab. 1 - Chemical analyses.

	(1) pitiglianoite (this work)	(2) pitiglianoite (Merlino et al., 1991)	(3) microsommite (Bonaccorsi et al., in prep.)
SiO_2	35.25	34.99	31.44
Al_2O_3	28.48	29.05	26.37
CaO	0.48	0.07	10.24
Na ₂ O	13.93	17.10	11.37
K_2O	12.13	9.41	6.98
SO ₃	6.20	7.58	6.41
Cl	0.09	0.01	6.92
H_2O	3.44*	3.46**	-
Σ '	100.00	101.67	99.73
O = Cl			1.56
Σ			98.17
Si	6.15	6.07	6.03
Al	5.85	5.93	5.97
Ca	0.09	0.01	2.11
Na	4.71	5.75	4.23
K	2.70	2.08	1.71
S	0.81	0.99	0.92
Cl	0.03	0.003	2.25
H ₂ O	2.00^{*}	2.00**	-

- (1) SEM/EDS analysis, average of five points. (*)The $\rm H_2O$ wt.% is calculated on the basis of two $\rm H_2O$ molecules $\it p.f.u.$ and the sum of oxides is normalized to 100.
- (2) Microprobe analysis, average of seven points. (**)The $\rm H_2O$ wt.% is calculated on the basis of two $\rm H_2O$ molecules $\it p.f.u.$
- (3) Microprobe analysis, average of seven points.

scanning electron microscope, working at 20 kV, 2500 cps, counting time of 100 s, spot size $200 \div 500 \text{ nm}$. The raw data were corrected by ZAF procedure. The H_2O weight % was calculated on the basis of two water molecules p.f.u., as obtained from structural refinement (Merlino et al. 1991).

The resulting chemical formula, on the basis of 12

(Si+Al), is:

Na_{4.7}K_{2.7}Ca_{0.1}Si_{6.1}Al_{5.9}O₂₄(SO₄)_{0.8}•2H₂O showing a content in potassium cations slighly higher than pitiglianoite from the type locality. An idealized formula for our sample is:

 $[Na_3K_3(SO_4)][Na_2(H_2O)_2][Si_6Al_6O_{24}^1]$

where the chemical composition of the three distinct structural modules - large channels, undecahedral cages and framework, respectively - is shown.

X-RAY STUDY

A single crystal was selected for X-ray characterization. By means of the oscillating-crystal and Weissenberg techniques, we obtained $c \sim 5.2$ and $a \sim 22.2$

Tab.2 - Cell parameters.

£-	pitiglianoite (this work)	pitiglianoite (Merlino et al.,1991)	microsommite (Bonaccorsi et al., in prep.)
a (Å) c (Å) V (ų)	22.230 (1) 5.2679 (5) 2254.4 (3)	22.121 (3) 5.221 (1) 2212.8 (6)	22.160 (2) 5.346 (1) 2273.5 (5)
space group	$P6_3 \text{ (or } P6_3/m)$	$P6_3$	P6 ₃

Å, Laue symmetry 6/m. In particular, (hk0) Weissenberg photographs clearly indicated the presence of a superstructure, by showing strong reflections with h - k = 3n, and weaker reflections with $h - k \neq 3n$ which pointed to a supercell parameter $a \sim 22.2$ A.

The cell parameters were derived through least-square fit of the 2θ values of 30 reflections ($25^{\circ} < 2\theta < 50^{\circ}$) measured on a Siemens P4 automatic four-circle diffractometer, by using a graphite monocromatized MoK α radiation. The cell parameters are a = 22.230(1), c = 5.2679(5) Å; they are larger than those of pitiglianoite from the type locality, in accordance with the higher content in potassium cation of our crystals. In Table 2 the cell parameters are compared with those of similar phases.

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